Structure of Organic Thin Films Grown on Surface-modified Tantalum Oxide

Ryo Onoki,1 Shigeomi Abe,1 Keiji Ueno,*1 Hiroo Nakahara,1 and Koichiro Saiki2

¹Department of Chemistry, Saitama University, 255 Shimo-Okubo, Sakura-ku, Saitama 338-8570

²Department of Complexity Science and Engineering, The University of Tokyo, 5-1-5 Kashiwanoha, Kashiwa 277-8561

(Received February 28, 2006; CL-060242; E-mail: kei@chem.saitama-u.ac.jp)

We fabricated an ordered monolayer film of a long-chain fatty acid or a perfluorofatty acid onto an anodized tantalum oxide (Ta_2O_5) amorphous surface by the Langmuir–Blodgett (LB) method, and grew a pentacene or C_{60} thin film on it. Remarkable improvements of morphology and crystallinity of the organic thin films were observed on the CH₃-terminated monolayer film of the fatty acid.

Organic semiconductors have drawn much attention due to their electronic functionality and mechanical flexibility. Development of organic field effect transistors (OFETs) using such soft materials is of great importance and its investigation has spread world-wide.^{1,2} In most of such OFETs, thermally oxidized SiO_x films on conductive Si substrates have been used as the gate dielectric. However, the dielectric constant (*k*) of the SiO_x layer is rather low so that it is difficult to inject a large amount of carriers into the organic layer with low gate voltages. Recently, many groups have reported the characteristics of the OFETs which were fabricated on high-*k* gate insulators.^{2,3} We have also succeeded in fabricating OFET using an anodized tantalum pentaoxide (Ta₂O₅) layer as the high-*k* gate dielectric.⁴

Another problem still remains if we use the high-k gate dielectric. The surface of the oxide layer is amorphous, so that a single-crystalline organic thin film never grows on the anodized Ta₂O₅ surface. It is reported that the performance of OFETs is mainly affected by the quality of the initial organic layer on the gate dielectric.^{5,6} Therefore, the improvement in the crystallinity of the dielectric surface is expected to reduce the defects or the domain boundaries in the organic layer, and improve the working characteristics of OFETs. As for the SiO_x surface, organosilane self-assembled monolayers (SAMs) are used to introduce the ordered structure, and improvement in the OFET performance has been reported.⁷⁻¹⁰ To our knowledge, however, no SAMs onto the Ta₂O₅ surface has been reported so far. In the present study, we focused our attention on the modification of the amorphous Ta₂O₅ surface by the LB films of long-chain fatty acids, which provides an artificial two-dimensional crystalline film on the solid substrates.¹¹ Then, we grew pentacene and C_{60} thin films on the modified Ta_2O_5 surfaces, and investigated the crystallinity of the grown films.

A 0.1 mm-thick Ta sheet was purchased from Furuuchi Chemical Corporation. A mirror-like anodized Ta₂O₅ surface was obtained following the previously reported processes.⁴ The Ta substrate cut from the sheet was electrolytically polished in an aqueous HF solution (50%):concentrated H₂SO₄ = 1:9 (v/v) mixed electrolytic solution. The polished substrate was anodized in a KI solution (2×10^{-4} mol·L⁻¹) at 300 V, which produced a 600 nm-thick anodized Ta₂O₅ layer on Ta. Octadecanoic acid (C₁₈A, Tokyo Chemical Industry Co., Ltd.) and perfluorooctadecanoic acid (PFC₁₈A, Fluorochem Ltd.) were pu-

rified by repeated recrystallization from hexane or chloroform, respectively. C₁₈A was dissolved in benzene and PFC₁₈A was dissolved in a trifluoroacetic acid:chloroform = 1:4 (v/v) solution to make the spreading solutions with the concentration of 1×10^{-4} mol·L⁻¹. Their monolayers were spread onto an aqueous subphase (pH = 6.8) containing CdCl₂ (3×10^{-4} mol·L⁻¹) and KHCO₃ $(5 \times 10^{-5} \text{ mol} \cdot \text{L}^{-1})$ in Milli-Q water (R = 18.2M Ω). Monolayers in the form of the cadmium salts (CdC₁₈ and CdPFC₁₈) were deposited by the LB method on the Ta₂O₅/Ta substrates at 15 °C with the surface pressure of $25 \text{ mN} \cdot \text{m}^{-1}$ using a Lauda FW-1 film balance. The k of the dielectric layer was only slightly reduced (less than several %) by the deposition. Pentacene (Aldrich, 98%) or C_{60} (99.99%) molecules were evaporated from a Knudsen cell onto the Ta₂O₅/Ta substrate at room temperature under the ultrahigh vacuum (ca. 10^{-7} Pa). The evaporation rate and the final film thickness were set to 1.0 nm·min⁻¹ and 100 nm, respectively. The wettability of the bare and modified Ta₂O₅ surfaces was observed by measuring the water contact angle. Surface morphology and crystallinity were investigated by atomic force microscopy (AFM) and X-ray diffraction (XRD).

The anodized Ta₂O₅ surface was smooth, and the rootmean-square roughness value of it was below 0.2 nm, which was as small as that of the thermally oxidized SiO_x. The anodized surface showed hydrophilicity with the water contact angle of 29°, and it was expected that the LB monolayer exposing the outermost layer of the hydrophobic group would be easily deposited on it. Actually, CdC18 and CdPFC18 monolayers were deposited with good transfer ratios (almost unity). The water contact angles were 103 and 104° for the CdC₁₈ and CdPFC₁₈ monolayers, respectively, and hydrophobic nature of these modified surfaces was confirmed. The values of contact angles agree well with the previously reported ones.^{12,13} AFM observation also showed very flat surfaces of the deposited monolayer films. Thus, we were able to obtain CH₃- or CF₃-terminated well-ordered Ta₂O₅ surfaces by the LB method. We think that the LB monolayer did not change the structure during the growth of pentacene and C₆₀ films, because they were grown at room temperature.

Figure 1 shows AFM images of the pentacene and C_{60} thin films on the above-mentioned surfaces. On the bare Ta₂O₅ surface, the pentacene film had very large domains with a dendritic shape similar to that on the SiO_x surface, and the layer-by-layer growth was clearly observed. The domain size of the C₆₀ film, however, was far smaller than that of the pentacene film. The nucleation density was very high, and there were quite a number of domain boundaries. It was not expected to grow a C₆₀ film with good crystallinity on the bare surface. On the CH₃-terminated Ta₂O₅ surface, we observed large domains of pentacene, which had the similar size with those on the bare Ta₂O₅ surface. It seemed that the nucleation density was a little smaller, but the



Figure 1. AFM images of the organic thin films. Pentacene and C_{60} films were grown on (a and d) bare Ta₂O₅ surfaces, (b and e) CdC₁₈ LB film surfaces, and (c and f) CdPFC₁₈ LB film surfaces, respectively.



Figure 2. XRD patterns of pentacene thin films grown on the bare Ta_2O_5 surface and the LB film surfaces.

change was not so dramatic. On the other hand, the morphology of the C_{60} film greatly changed on the CH₃-terminated surface. Larger C_{60} domains with lower nuclear density were observed in the AFM image as compared with the film on the bare Ta₂O₅ surface. Therefore, it was suggested that the well-ordered C_{60} thin film grew on the CH₃-surface. On the CF₃-terminated surfaces, both pentacene and C_{60} films formed far smaller domains than on the CH₃-terminated surfaces. Especially, there were a very large number of domain boundaries in the pentacene film.

Figure 2 shows XRD patterns of the pentacene thin films on the various surfaces. Diffraction peaks assigned to (001), (002), (003), and (004) were found in each pattern. Two phases coexisted in the pentacene films on the bare and CF3-terminated surfaces. They correspond to the thin film phase and the bulk phase with interplanar spacing (d) of 1.5 and 1.4 nm for the (001) reflection, respectively.¹⁴ For the film on the CF₃-terminated surface, the peak intensities were weakened and the bulk phase was dominant. In contrast, the pentacene film consisted of only the thin film phase on the CH3-terminated surface. Figure 3 shows XRD patterns of the C₆₀ films. No diffraction peak was observed on the bare and CF3-terminated surfaces, suggesting the growth of amorphous films. On the CH₃-terminated surface, however, there appeared a sharp peak with the corresponding d spacing of 0.8 nm for the (111) reflection, and an additional (222) reflection peak, indicating the growth of a highly-oriented C₆₀ film.

These results imply that the *c*-axis ordering of the pentacene and C_{60} thin films was greatly improved by fabricating on the CdC₁₈ LB films. The CH₃-terminated surface shows the lipophilicity, and the surface free energy is close to that of the organic molecules. So both pentacene and C_{60} molecules are enough wettable with the CH₃-terminated surface. Then, the formation of larger domains is favored at the initial growth stage on the



Figure 3. XRD patterns of C_{60} thin films grown on the bare Ta_2O_5 surface and the LB film surfaces.

CH₃-terminated surface due to the sufficient molecule–substrate interaction, and the crystallinity of the organic thin films can be improved.

In the case of the bare Ta_2O_5 surface, it has some lipophobicity so that C_{60} molecules, which have no C–H bond, are not enough wettable and form an amorphous film. In contrast, pentacene molecules are wettable on Ta_2O_5 and form highly *c*-axisoriented layer-by-layer domains. Meanwhile, the CF₃-terminated surface provides the lowest surface free energy with the higher water and oil resistances. On it, the intermolecular interaction dominates over the molecule–substrate interaction. Then, many small domains are formed at the initial stage of the growth, because most impinging molecules nucleate not at the side step edges of preexisting domains, but just onto the domain surfaces. Thus, aggregated small domains with a large number of domain boundaries are formed even for pentacene molecules.

In conclusion, we succeeded in introducing ordered structures onto amorphous Ta_2O_5 surfaces by the LB method. The crystallinity of organic films, especially of the C_{60} film, was greatly improved by the surface modification with the CdC₁₈ film. This method is effective for the improvement of the high-*k* gate surfaces, on which good SAMs could not be grown due to the lack of appropriate self-assembling molecules.

This work was supported by the Grant-in-Aid for Creative Scientific Research from the Ministry of Education, Culture, Sports, Science and Technology of Japan (No. 14GS0207).

References

- T. Oyamada, H. Sasabe, C. Adachi, S. Okuyama, N. Shimoji, K. Matsushige, *Appl. Phys. Lett.* 2005, 86, 093505.
- 2 Y. Iino, Y. Inoue, Y. Fujisaki, H. Fujikake, H. Sato, M. Kawakita, S. Tokito, H. Kikuchi, Jpn. J. Appl. Phys. 2003, 42, 299.
- 3 C. D. Dimitrakopoulos, S. Purushothaman, J. Kymissis, A. Callegari, J. M. Shaw, *Science* 1999, 283, 822.
- K. Ueno, S. Abe, R. Onoki, K. Saiki, J. Appl. Phys. 2005, 98, 114503.
 M. Kiguchi, M. Nakavama, K. Fujiwara, K. Ueno, T. Shimada, K. Saiki,
- 5 M. Kiguchi, M. Nakayama, K. Fujiwara, K. Ueno, T. Shimada, K. Saiki, *Jpn. J. Appl. Phys.* **2003**, *42*, L1408.
- 6 M. Kiguchi, M. Nakayama, T. Shimada, K. Saiki, *Phys. Rev. B* 2005, 71, 035332.
- 7 E. E. Polymeropoulos, J. Sagiv, J. Chem. Phys. 1978, 69, 1836.
- 8 R. D. Peters, P. F. Nealey, J. N. Crain, F. J. Himpsel, Langmuir 2002, 18,
- 1250.
 9 Y.-Y. Lin, D. J. Gundlach, S. F. Nelson, T. N. Jackson, *IEEE Electron Device Lett.* 1997, *18*, 606.
- 10 S. Kobayashi, T. Nishikawa, T. Takenobu, S. Mori, T. Shimoda, T. Mitani, H. Shimotani, N. Yoshimoto, S. Ogawa, Y. Iwasa, *Nat. Mater.* 2004, *3*, 317.
- M. C. Petty, *Langmuir-Blodgett Films*, Cambridge University Press, Cambridge, 1996.
- E. F. Hare, E. G. Shafrin, W. A. Zisman, J. Phys. Chem. 1954, 58, 236.
 E. Györvary, J. Peltonen, M. Lindén, J. B. Rosenhom, Thin Solid Films
- 1996, 284–285, 368.
 C. D. Dimitrakopoulos, A. R. Brown, A. Pomp, J. Appl. Phys. 1996, 80, 2501.